

CHEMICAL HAZARDS AND PERSONAL PROTECTIVE EQUIPMENTS

Editor: Chen-Peng Chen, Ph.D.

Department of Occupational Safety and Health

China Medical University

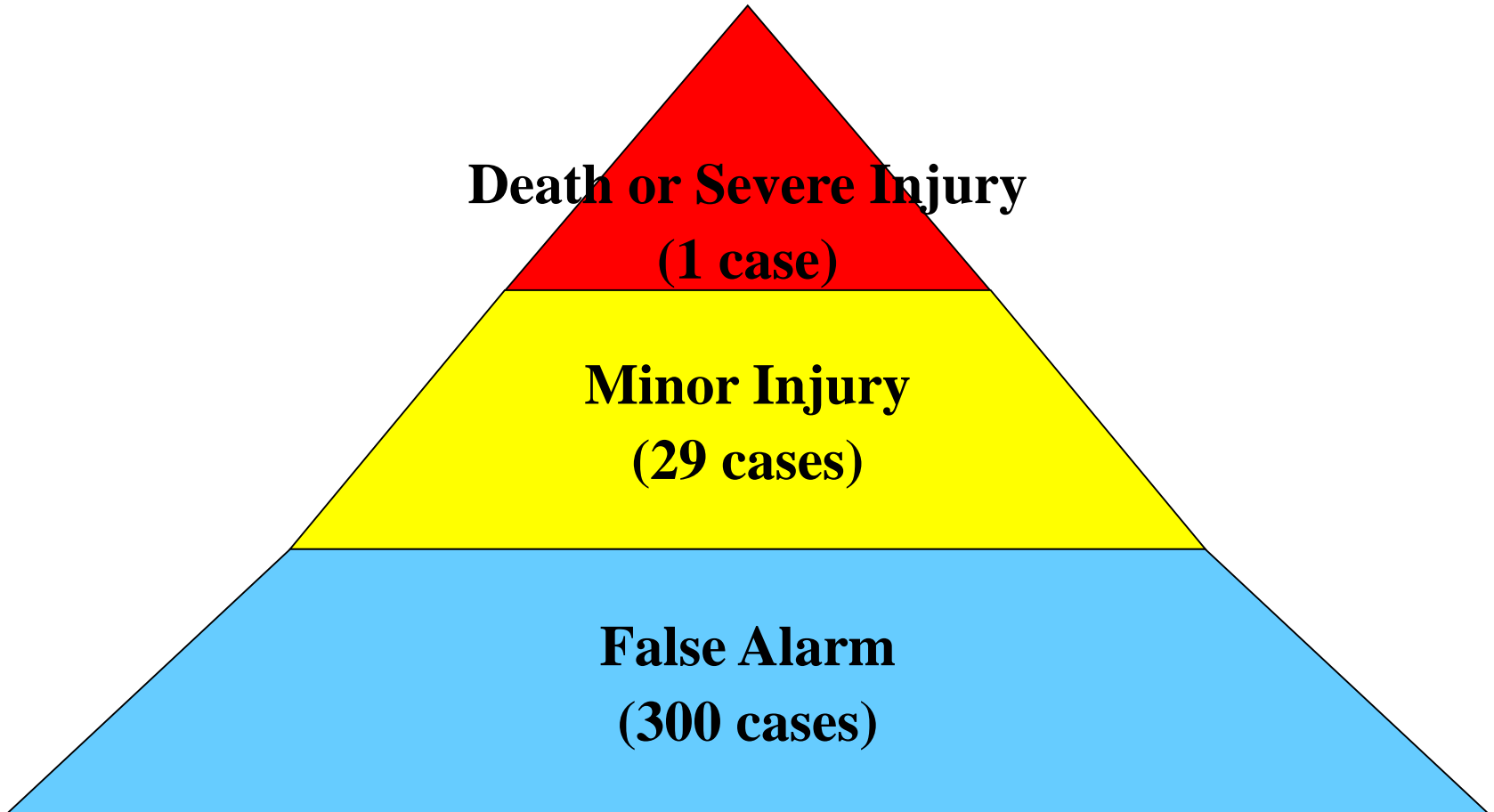
Learning Goals

- Understanding chemical hazards present in the laboratory and strategy of prevention and control
 - Hazard communication for chemical substances
 - Pointers on management of chemical hazards
- Understanding different types of personal protective equipments (PPEs) commonly used in the laboratory and workplace
- Learning how to properly select, use, maintain, and store PPEs

Characteristics of Laboratory as a Workplace

- Presence of miscellaneous hazardous, harmful, or toxic chemicals in significant quantity
- Change in personnel by rotation or entry of new employees at a significant level
- Elevated risk of unknown and uncertain nature owing to development of new technique and research
- High density of technology and equipment

The Risk Pyramid



Significance of Unsafe Behavior and Status

Outlines

Section 1: Chemical Hazards and Strategy of
Evaluation and Control

Section 2: Hazard Communication for Hazardous
Chemical Substances

Section 3: Pointers on Management of Chemical
Hazards

Section 4: Overview of Personal Protective
Equipments and Eye/Facial Protection

Section 6: Respirators

Section 7: Protective Clothing and Hand Protection

Section 1

Chemical Hazards and Strategy of Evaluation and Control

Definition of Hazard

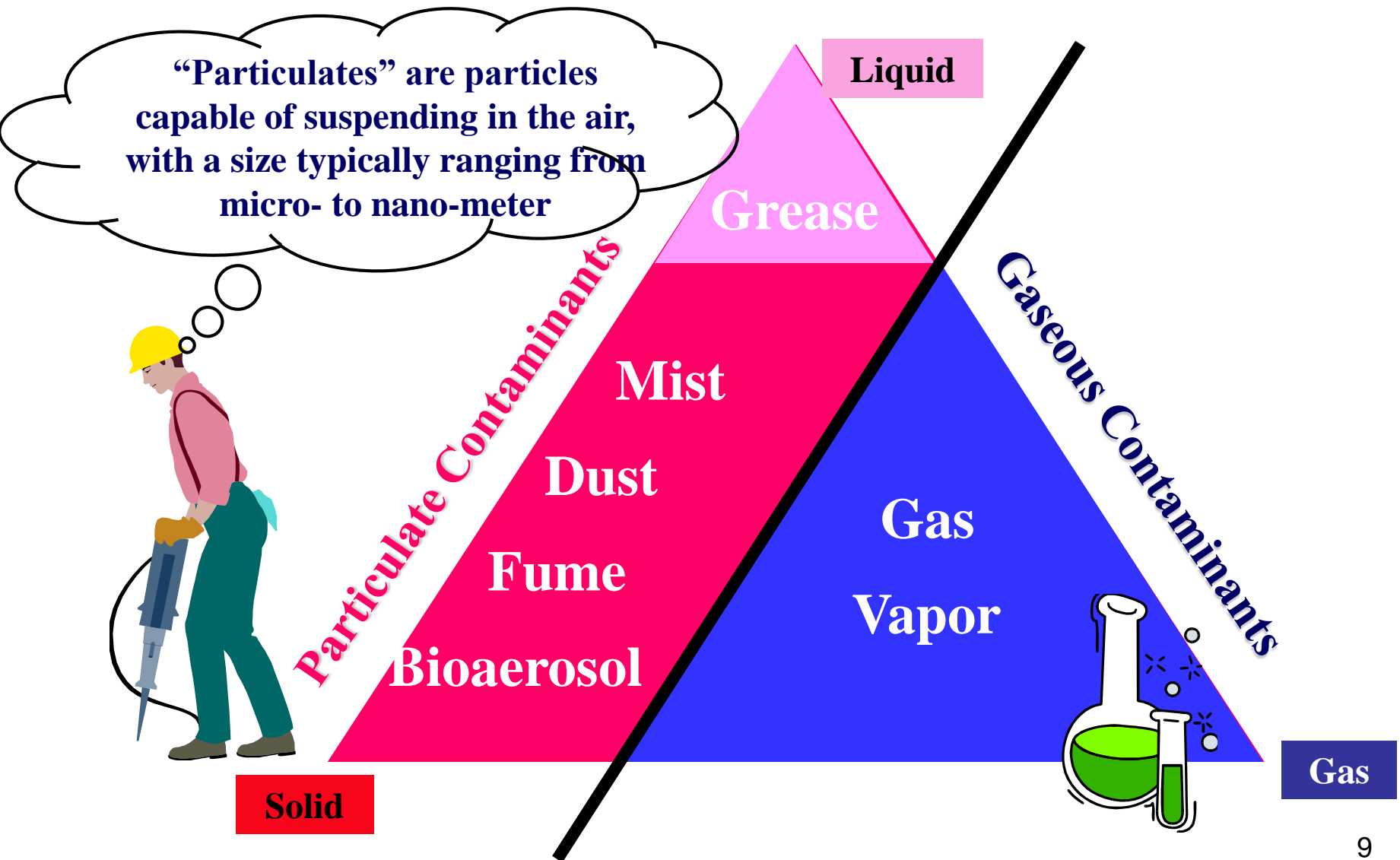
- Hazard: any sources of potential damage, harm or adverse health effects on something or someone under certain conditions
 - A hazard can cause harm or adverse effects to individuals as health effects, or to organizations as property or equipment losses
 - There are five major categories of hazard present in the workplace: chemical, physical, biological, ergonomic and safety hazards

Chemical Hazards

- Toxicity: harmful effect such as poisoning or tissue ulceration due to *contact or uptake of chemical* by route of inhalation, ingestion, or skin contact
- Hazard: the damage caused by chemical due to *energy released from chemical reaction*, such as fires and explosions



Classification of Airborne Hazards



Classification of Airborne Hazards

- The airborne hazards can be in the form of solid, liquid, or gas/vapor:
 - Dusts are solid particles suspending in the air; they can be originating from cutting, grinding, or drilling of solid material, and in general are of a size greater than 5 μm
 - Examples include lead dusts present in battery-manufacturing facilities
 - Fumes are solid particles made of metals suspending in the air due to heating and subsequent condensation; in general the particles have a size less than 0.5 μm
 - Examples include tin/manganese fumes generated from welding works

Classification of Airborne Hazards

- A third class of dusts receiving great attention are biological aerosols, such as those of bacteria, viruses, and fungi
 - Some *bioaerosols* are capable of entering the lungs at a great depth when inhaled
- Liquid particulates are generally referred to as mists
 - Examples include those formed from droplets of lubricating oil used in metal-cutting work
 - If the originating solution is acidic, then the mists are called *acidic mists*, such as sulfuric acid mists or chromic acid mists; both acid mists are highly corrosive

Classification of Airborne Hazards

- Gas molecules possess a significant tendency to diffuse, and they evenly occupy the diffused space
 - Examples include carbon dioxide, ethene (used in welding), carbon monoxide from combustion, and etc
- Vapor consists of molecules evaporating from liquid or subliming from solid due to heating or provision of energy; these molecules otherwise stay as liquid or solid under normal temperature and pressure
 - Examples include organic solvents used in paints or other commonly known solvents such as benzene, n-hexane, carbon tetrachloride, all have been reported in industrial accidents and some confirmed as carcinogens

Case study: Fire from flammable solvent destroying laboratory

- A glass container carrying 4 liters of n-hexane ruptured in a university's laboratory
- The chemical might have evaporated when the graduate student mopped off the spill and subsequently came in contact with the temperature control of a furnace, resulting in a fire
- The fire was put off after two and a half hours; three laboratories were directly destroyed, with a loss of approximately 10 million NTD

Hazard Evaluation and Control

- Hazard evaluation
 - A process to determine type and severity of hazard based on the work/experimental process and the equipment/material used in the process
 - For chemical hazard, the evaluation includes review of chemical's **Material Safety Data Sheet** to understand physicochemical characteristics of the substance in the environment (e.g. solid/liquid/gas or vapor; corrosivity; volatility; and lipophilicity) as well as its toxicity

Hazard Evaluation and Control

- Priority of hazard prevention

- 1) **Engineering control**

- ***Substitution***: Replacing highly toxic stock material with those of lower toxicity; substituting high-risk experimental process with low-risk one
 - ***Reduction***: Experimenting with less stock material
 - ***Isolation***: Separating lab workers not directly involved in handling hazard-producing equipment or experimental process from the work zone
 - ***Ventilation***: Local exhaust ventilation or general ventilation (dilute ventilation)

Hazard Evaluation and Control

- Priority of hazard prevention

2) Administrative control

- Education and training; personal hygiene; workplace cleaning and maintenance; material labeling; worker rotation; environmental monitoring; and health surveillance

3) PPEs

- These are not an alternative to engineering and administrative controls
- They should be used only as *the last line of defense*
- Occupational emphasis has been on respirators and chemical protective clothing (including gloves)

Section 2

Hazard Communication for Hazardous Chemical Substances

Definition of Hazard Communication

Communication on potential hazards:

- Workplace hazards may be better controlled or prevented with increased awareness and protective measures
- Major contents in hazard communication
 - Characteristics of potential hazards
 - Prevention and control measures
 - Emergency response procedures

Hazard communication may be used in many types of work environments...



In this session, we will focus on hazard communication for chemicals only

Labeling in Hazard Communication

- Hazard Labeling
 - Graphic presentation and cautions (words/phrases) according to major hazards
- Safety Data Sheet (SDS)
 - Providing information on chemical hazards and safety recommendations
- Most of the labeling and communication requirements now follow the *GHS* guidelines

Globally Harmonized System of Classification and Labelling of Chemicals (GHS)

- Providing general guideline on chemical hazard classification and labeling
 - Internationally harmonized system on *hazard classification*
 - Uniform format on hazard identification and safety message
 - Uniform format on hazard information and recommended safety measures

GHS Classification for Chemical Hazards










29 GHS hazard classes...

- Classification based on primary hazards and physical state of the chemical (solid, liquid or gas)
- Two or more categories are given within each hazard class
- The hazards classes are also characterized into three major groups:
 - *Physical hazards*
 - *Health hazards*
 - *Environmental hazards*

Hazard Labeling Requirements

- Current regulatory requirements on hazard labeling follows the GHS format and include:
 - Pictograms for major hazards
 - Name of substance (or ingredient)
 - Signal word
 - Hazard statements
 - Precautionary statements
 - Supplier identification

Pictograms for Chemical Hazards

Flammable	Oxidizing	Explosive
		
Corrosive	Pressurized gas	Acute toxicity
		
Health hazard	Health hazard	Environmental hazard
		

Chemicals with Health Hazards

Two symbols for health hazards

- Respiratory sensitization
- Mutagenicity
- Carcinogenicity
- Reproductive toxicity
- Toxicity to target organ from single or long-term exposure
- Aspiration



Acute toxicity



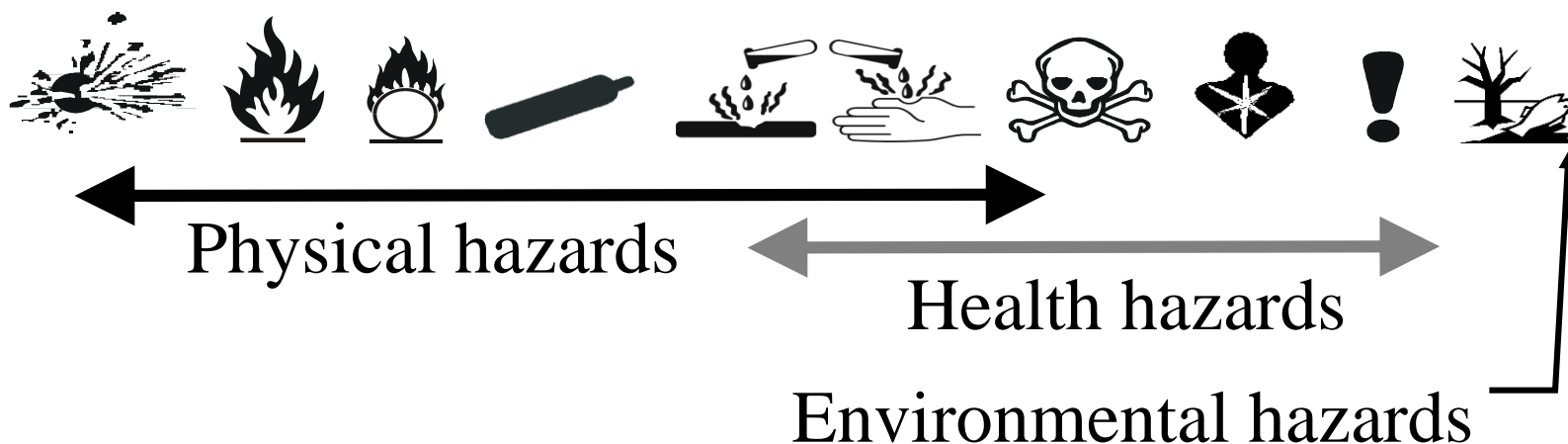
Irritant to skin
and/or eyes



Skin sensitization



Pictograms for Chemical Hazards



*Not all chemical hazards are assigned pictograms;
some chemicals have more than one pictogram*



Other GHS Label Elements

- Signal word
 - Single-word identifier for hazard: “Danger” or “warning”
- Hazard statement
 - Specification for type of hazard and conditions potentially leading to hazard
 - Phrases with hazard category
- Precautionary statement
 - Summary for safe handling and storage of the substance
- Supplier identification
 - Listing of name, address and phone number of chemical manufacturer or supplier

Labeling for Chemical Hazards

- For dangerous and harmful materials, a hazard label is required outside of the container/package, outside the primary storage area, and nearby major area of usage
 - Pictograms must be displayed whenever applicable
 - For shipping packages, additional requirements on transportation labeling must also be met

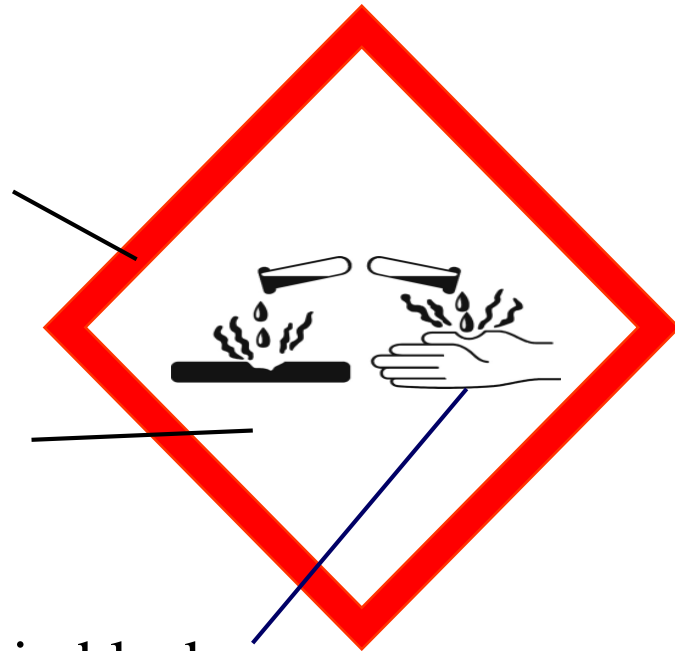
Posting of Sign for Chemical Hazards

- Signs for hazard communication must be posted outside of major storage and handling area at locations that may be easily noticed

Diamond shape,
min. 10 cm × 10 cm

White background
with red border

Hazard pictogram in black



Example of Chemical Hazard Label

Chemical/ingredient name

Hazard pictograms

Signal word: “Danger” or “Warning”

Hazard statements

Precautionary statements

Supplier identification

TOLUENE
CAS # 108-88-3.

Obtain special instructions before use. Do not handle until all safety precautions have been read and understood.

DANGER

Highly flammable liquid and vapour. May be fatal if swallowed and enters airways. Causes damage to central nervous system if inhaled. Causes damage to central nervous system, liver and kidneys through prolonged or repeated exposure. May damage fertility or the unborn child. Harmful if inhaled. May be harmful if swallowed. Causes skin irritation. May cause respiratory irritation. May cause drowsiness and dizziness. Toxic to aquatic life.

Keep away from heat, sparks and flame - No smoking. Take precautionary measures against static discharge. Ground/bond container and receiving equipment. Use only non-sparking tools. Do not breathe vapours. Wear protective gloves and eye/face protection. Use only in a well-ventilated area. Keep container closed when not in use. Store in a cool, well-ventilated place away from heat and ignition sources. Store locked up in a closed container.
IN CASE OF FIRE: Use carbon dioxide, dry chemicals or appropriate foam.

FIRST AID:
IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician. Do NOT induce vomiting. IF ON SKIN: Remove/take off immediately all contaminated clothing. Wash with plenty of soap and water. Get medical advice/attention. IF INHALED: Call a POISON CENTER or doctor/physician.

REFER TO SAFETY DATA SHEET

 NIAGARA FALLS NEW YORK HOUSTON TEXAS MISSISSAUGA ONTARIO DORVAL QUEBEC

Example of Chemical Hazard Label

- Additional transportation label may be required during shipment

Transportation label

The diagram shows a vertical rectangular container with horizontal bands. A red diamond-shaped hazard label is affixed to the front, featuring a flame symbol and the number '3'. A callout box labeled 'Transportation label' points to this diamond. Another callout box points to a rectangular area below the diamond, indicating where an additional transportation label might be required. To the right, a detailed hazard label for Toluene is shown, which includes the product name, CAS number, hazard pictograms, and detailed safety information.

TOLUENE
CAS # 108-88-3.

Obtain special instructions before use. Do not handle until all safety precautions have been read and understood.

3

DANGER
Highly flammable liquid and vapour. May be fatal if swallowed and enters airways. Causes damage to central nervous system if inhaled. Causes damage to central nervous system, liver and kidneys through prolonged or repeated exposure. May damage fertility or the unborn child. Harmful if inhaled. May be harmful if swallowed. Causes skin irritation. May cause respiratory irritation. May cause drowsiness and dizziness. Toxic to aquatic life.

Keep away from heat, sparks and flame - No smoking. Take precautionary measures against static discharge. Ground/bond container and receiving equipment. Use only non-sparking tools. Do not breathe vapours. Wear protective gloves and eye/face protection. Use only in a well-ventilated area. Keep container closed when not in use. Store in a cool, well-ventilated place away from heat and ignition sources. Store locked up in a closed container.
IN CASE OF FIRE: Use carbon dioxide, dry chemicals or appropriate foam.

FIRST AID:
IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician. Do NOT induce vomiting. IF ON SKIN: Remove/take off immediately all contaminated clothing. Wash with plenty of soap and water. Get medical advice/attention. IF INHALED: Call a POISON CENTER or doctor physician.

REFER TO SAFETY DATA SHEET

The Compliance Center Inc.™
NIAGARA FALLS NEW YORK HOUSTON TEXAS MISSISSAUGA ONTARIO DORVAL QUEBEC

Safety Data Sheet

- To provide detailed health and safety information on potential hazards, emergency response procedures, safe storage, handling and disposal of the chemical
- Current requirements for safety data sheet (SDS) also follow GHS specifications
 - SDS is a format of MSDS in consistence with GHS guideline; as far as chemical management in laboratory is concerned, these two terms are interchangeable

A Complete SDS Contains 16 Sections...

1. Product and Company identification
2. Hazard identification
3. Composition/ingredient
4. First-aid measures
5. Fire-fighting measures
6. Accidental release measures
7. Handling and Storage
8. Exposure control/personal protection

A Complete SDS Contains 16 Sections...

9. Physical and chemical properties
10. Stability and reactivity
11. Toxicological information
12. Ecological information
13. Disposal considerations
14. Transport information
15. Regulatory information
16. Other information

Safety Data Sheet (SDS)

- SDS are required for display in laboratory handling chemical, and should be displayed at visible locations of easy access
- Contents of SDS in current display should be confirmed for accuracy and updated periodically
 - The updated record should be on file for *three years*
- Sections of SDS directly relevant to emergency response: *First Aid Measures, Fire-Fighting Measures, Accidental Release Measures, and Exposure Controls/Personal Protection*

Ben		A SDS for benzene									
Version 1.6		Revision Date 2011-08-24									
1. IDENTIFICATION OF THE SUBSTANCE/MIXTURE AND OF THE COMPANY/UNDERTAKING											
Product information											
Trade name	: Benzene										
Material	: 1098293, 1059192, 1059060, 1037212, 1037213, 1037103, 1029170, 1037104, 1015526, 1016960										
Company	: Chevron Phillips Chemical Company LP 10001 Six Pines Drive The Woodlands, TX 77380										
Emergency telephone:											
Health: 866.442.9628 (North America) 1.832.813.4984 (International)											
Transport: North America: CHEMTREC 800.424.9300 or 703.527.3887 Asia: +800 CHEMCALL (+800 2436 2255) China: 0532.8388.9090 EUROPE: BIG +32.14.584545 (phone) or +32.14583516 (telefax) Chemcare Asia: Tel: +65 6848 9048 - Mob: +65 8382 9188 - Fax: +65 6848 South America SOS-Cotec Inside Brazil: 0800.111.767 Outside Brazil: +55.19.3467.1600											
Responsible Department	: Product Safety and Toxicology Group										
E-mail address	: MSDS@CPChem.com										
Website	: www.CPChem.com										
2. HAZARDS IDENTIFICATION											
Emergency Overview											
<table border="1"> <tr> <td>Physical state: Liquid</td> <td>Color: Clear, colorless</td> <td>Odor: sweet, distinct</td> </tr> </table>				Physical state: Liquid	Color: Clear, colorless	Odor: sweet, distinct					
Physical state: Liquid	Color: Clear, colorless	Odor: sweet, distinct									
<table border="1"> <tr> <td>OSHA Hazards</td> <td>: Flammable Liquid, Carcinogen, Moderate skin irritant, Moderate eye irritant, Mutagen, Target Organ Effects</td> </tr> </table>				OSHA Hazards	: Flammable Liquid, Carcinogen, Moderate skin irritant, Moderate eye irritant, Mutagen, Target Organ Effects						
OSHA Hazards	: Flammable Liquid, Carcinogen, Moderate skin irritant, Moderate eye irritant, Mutagen, Target Organ Effects										
GHS Classification											
<table border="1"> <tr> <td>: Flammable liquids, Category 2</td> </tr> <tr> <td>: Skin irritation, Category 2</td> </tr> <tr> <td>: Eye irritation, Category 2A</td> </tr> <tr> <td>: Germ cell mutagenicity, Category 2</td> </tr> <tr> <td>: Carcinogenicity, Category 2</td> </tr> <tr> <td>: Aspiration hazard, Category 2</td> </tr> <tr> <td>: Acute aquatic toxicity, Category 2</td> </tr> <tr> <td>: Specific target organ systems - respiratory, Category 1, Inhalation</td> </tr> </table>				: Flammable liquids, Category 2	: Skin irritation, Category 2	: Eye irritation, Category 2A	: Germ cell mutagenicity, Category 2	: Carcinogenicity, Category 2	: Aspiration hazard, Category 2	: Acute aquatic toxicity, Category 2	: Specific target organ systems - respiratory, Category 1, Inhalation
: Flammable liquids, Category 2											
: Skin irritation, Category 2											
: Eye irritation, Category 2A											
: Germ cell mutagenicity, Category 2											
: Carcinogenicity, Category 2											
: Aspiration hazard, Category 2											
: Acute aquatic toxicity, Category 2											
: Specific target organ systems - respiratory, Category 1, Inhalation											

Example

(Source: Chevron Philips Chem Co;
http://www.cpchem.com/msds/100000068511_SDS_US_EN.PDF)

Section 3

Pointers on Management of Chemical Hazards

Strategy in Hazard Management

- Laboratory personnel should observe the following elements when working in laboratory to facilitate a proper evaluation of potential hazard and take preventive action
 - *Codes of safe laboratory practice*
 - *Material used in the experiment*
 - *Equipment/apparatus and protocol used in the experiment*
- Strategies of hazard prevention adopted in different types of laboratories might vary significantly, in the next slides only examples common to most laboratories are provided to illustrate cautions and strategies for hazard prevention

Storage of Chemical

- Hazardous material should be stored in accordance with its properties (e.g., *volatility*, *flammability*, and *compatibility*)
- The exhaust ventilation system installed in the storage facility should receive periodic inspection and maintenance
- Facilities storing a significant level of volatile, flammable liquid should be installed with *flammable gas detectors*

← **Exhaust ventilation**



Fire and explosion safety cabinet

***Fire and explosion safety cabinet
for storage of hydrogen cylinders***



Ventilation Equipments

- Volatile chemical should be processed and handled in a ventilation hood
- Samples containing microorganisms of airborne capability should be processed and handled in a biosafety cabinet
- Ventilation hood for chemical processing and biosafety cabinet for handling micro-organism are different in functions and the ways they function—don't mix-use them
- Do not leave material/tool in the hood unless necessary; they may influence stability of air flow for ventilation



Ventilation hood



Local exhaust ventilation

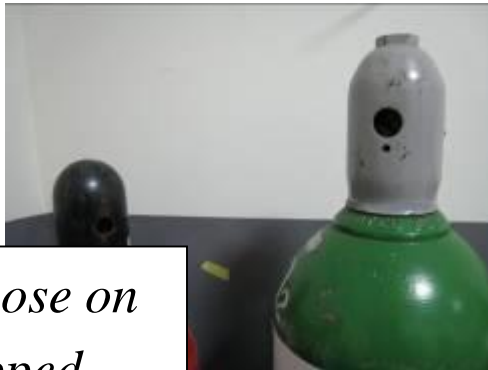
Ventilation Equipments

- If toxic gas may be released from scientific equipment during its operation, the point of gas emission should be covered in local exhaust ventilation
- Local exhaust ventilation, ventilation hood, and relevant equipments are required to be annually checked (e.g. for *transport velocity*)
- Stop your experiments immediately when the followings occur and seek help to check/repair:
 - Damage in exhaust pipeline
 - Abnormal operation of driving motor
 - Blocking of filtration device
 - Any other signs of abnormality (e.g. irregular sounds)

Gas Cylinders

- When using a high-pressure gas cylinder, observe the followings:
 - Cylinder transversely fixed?
 - All *gauge pressures* normal?
 - Inflammable objects present in the cylinder storage room?
 - Constituents of gas in cylinder clearly labeled?
 - Leakage from connection points?
 - Temperature in the storage room *over 40°C*?

Empty cylinders or those on reserve should be capped



Do not leave a wrench on cylinder-regulating valve!



Cylinder has to be tightly fixed!

Use and Management of Toxic Chemical Substances

- Containers and packaging of toxic chemical substances (TCS) should be labeled as required and attached with a copy of MSDS
- Entrance of facility where chemical is handled should be labeled with the sign “*Handling Premises of Toxic Chemicals*”
- Mechanism preventing release or leakage of TCS should be activated and operated when chemical is processed; emergency response equipment should be made available



Use and Management of Toxic Chemical Substances

- TCS should be stored in airtight, sturdy containers or packaging
 - Storage room or storage facility for TCS needs to be locked when the chemical are stored in the room
- Emergency response equipments and detecting/alarming devices should be periodically checked and calibrated
- The amounts of TCS in storage need to be in consistence with those on record
- The record of operation for TCS should remain on file for three years at the location of operation

Laboratory Waste

- Arbitrary disposal of laboratory waste including those of radioactivity, toxicity, corrosivity, inflammability, and infectious nature may pose a threat to health of those who work in the laboratory as well as to the environment
 - Arbitrary disposal of laboratory waste is subject to lawful punishment
- The collection, classification, labeling, and storage of laboratory waste and the transfer of waste to the management unit for temporary storage and final shipment for disposal should abide by the Institute's regulations



Section 4

Overview of Personal Protective Equipments and Eye/Facial Protection

Overview of PPEs

- **Types of PPEs for non-emergency use**

- ***Eye and facial protection***: Safety glasses and face shields
- ***Respirators***: Disposal dust masks and gas masks (for non-emergency use)
- ***Chemical protective clothing***: Lab coat and safety apron
- ***Hand and foot protection***: Gloves and safety shoes/boots
- ***Hearing protection***: Earplugs and earmuffs
-



Overview of PPEs

- **Types of PPEs for non-emergency use**

- PPEs can protect you only when you use them: Many PPEs were bought, displayed in the safety equipment cabinet, and never used
- PPEs may bring you more harm than safety when they are improperly used: PPEs have to fit the user's anthropometric characteristics such as the body shape, body size, and even facial characteristics in order to be protective
 - The risk from being exposed to toxic substance when using PPEs may be greater than when not using PPEs, e.g., dermal absorption of chemical is typically one-to-two magnitude greater than expected when the skin is moist

Laboratory Safety and Health Essentials...

➤ Safety glasses and lab coat



(<http://www.ecm.auckland.ac.nz/safety/safety.html>)

Laboratory Safety and Health Essentials...

➤ Closed shoes

 CLOSED SHOES				
 OPEN SHOES				

(<http://www.ecm.auckland.ac.nz/safety/safety.html>)

Eye and Facial Protection

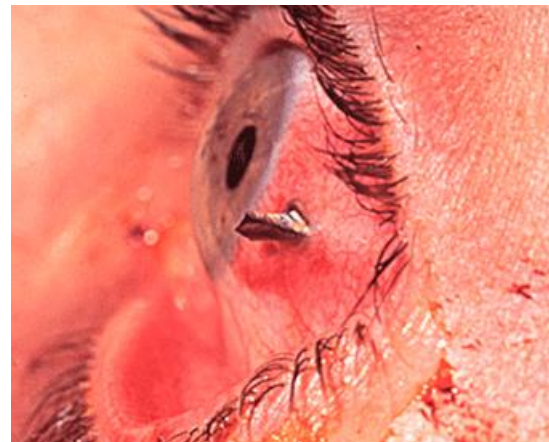
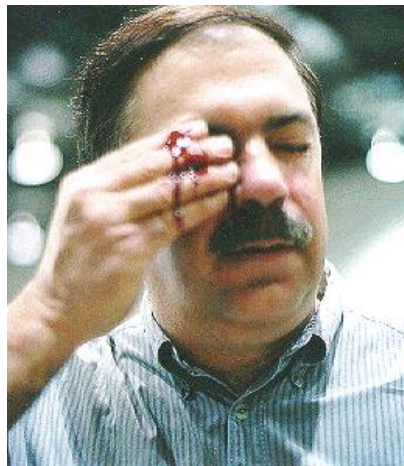
- *If these protective gears were not in place...*



This Pipefitter Escaped Serious Facial Injury By Wearing His Personal Protective Equipment



- *The consequences would be...*



Eye and Facial Protection

- *Factors prone to hurting your eyes and face in a laboratory*

Hazard Type	High-Risk Task
Impact	Cutting, slicing, grinding, sculpturing, and any other tasks producing objects or debris capable of flying with sufficient momentum
Heat	Operating high-temp furnace/oven, melting/welding work, and any other tasks producing high temp
Chemical	Handling and treatment of chemical or specimen
Dust	Woodwork
Light	Welding, melting and cutting, and laser operation

Eye and Facial Protection

- The protection offered by safety glasses
 - Mechanical harms, e.g. flying objects and spilled liquid
 - ✓ The shields on the side arms of safety glasses are designed to prevent entry of foreign objects
 - Radiation, e.g. ultraviolet and visible light radiation
 - ✓ Many facilities handling biotechnology use ultraviolet light as a tool for DNA observation; the ultraviolet light may cause damage to the cornea when a worker is continuously exposed to

Eye and Facial Protection

- Types of eye and facial protection

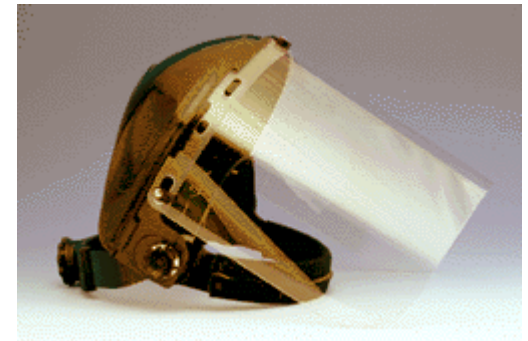
Safety Glasses



Goggles



Face Shields



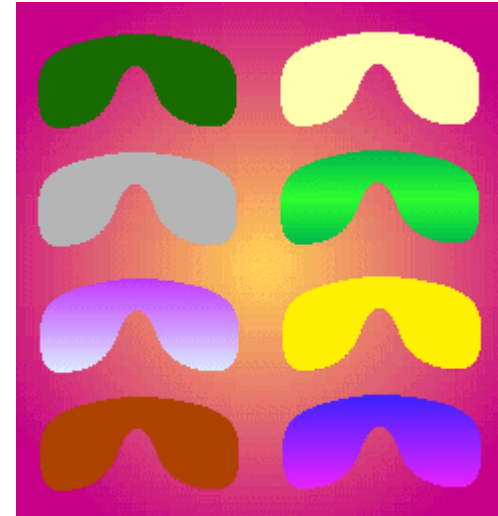
Eye and Facial Protection

- Types of eye and facial protection

Welding Face Shields and Helmets



Light-Filtering Lens



- Aiming to protect against light of high intensity

Eye and Facial Protection

- *Something about contact lenses*
 - In laboratory:
 - 1) Wearing contact lenses is prohibited (as the most conservative intervention approach)
 - 2) Evaluating risk of using contact lenses is a must-do
 - 3) Using airtight goggles for additional protection if you must use contact lenses
 - If by accident chemical is spilled in the eyes:
 - 1) Rinsing the eyes with clean water and removing contact lenses immediately
 - 2) Continuing to rinse for at least **15-30 minutes**
 - 3) Transferring the wounded to hospital

Eye and Facial Protection

- *Contact lenses are not intended for use as PPEs*

- Contact lenses are prohibited in the lab because:

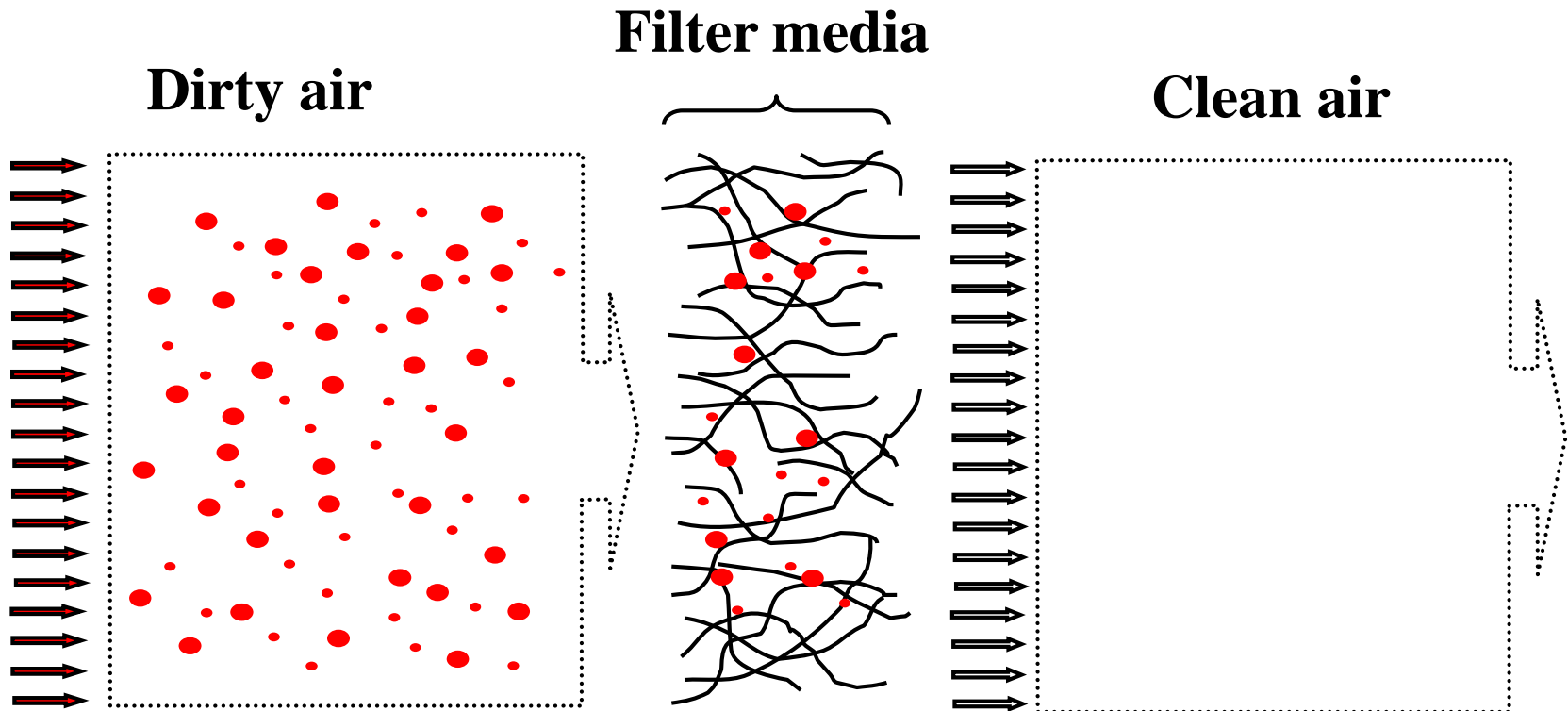
- 1) The reflective shutting of the eyelids when foreign objects enter makes removal of contact lenses difficult, unnecessarily delaying emergency treatment and medical attention required for the wounded
- 2) Some chemical in the form of vapor are capable of permeating through the lenses, and interacting extensively with ocular tissues due to occlusion effect
- 3) Long-term use of contact lenses may reduce sensitivity of cornea and secretion of tears, and thus weakening the eyes' natural defense

Section 5

Respirators

Respirators

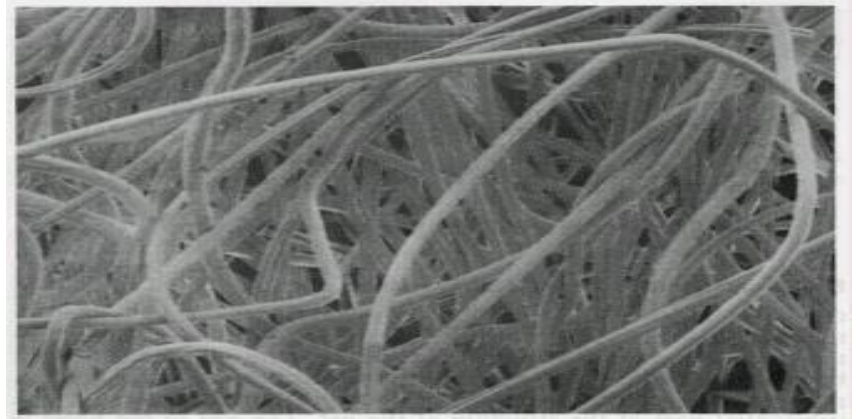
- “*Dust masks*”—removal of solid/liquid particulates



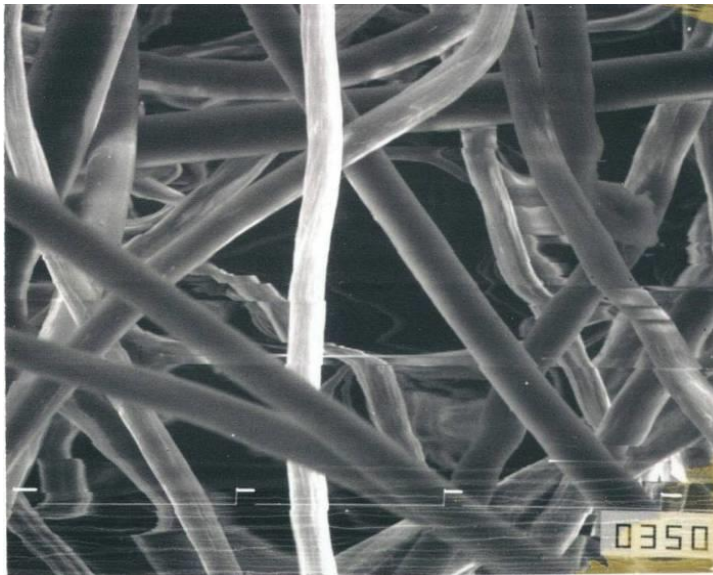
(Source: Organic solvent loading in canisters; Nelson, 1976)

Respirators

- *Filtration of particulates by fibers and aging effect for mechanical filters*



(Source: Air Filtration by R.C. Brown, Pergamon Press)



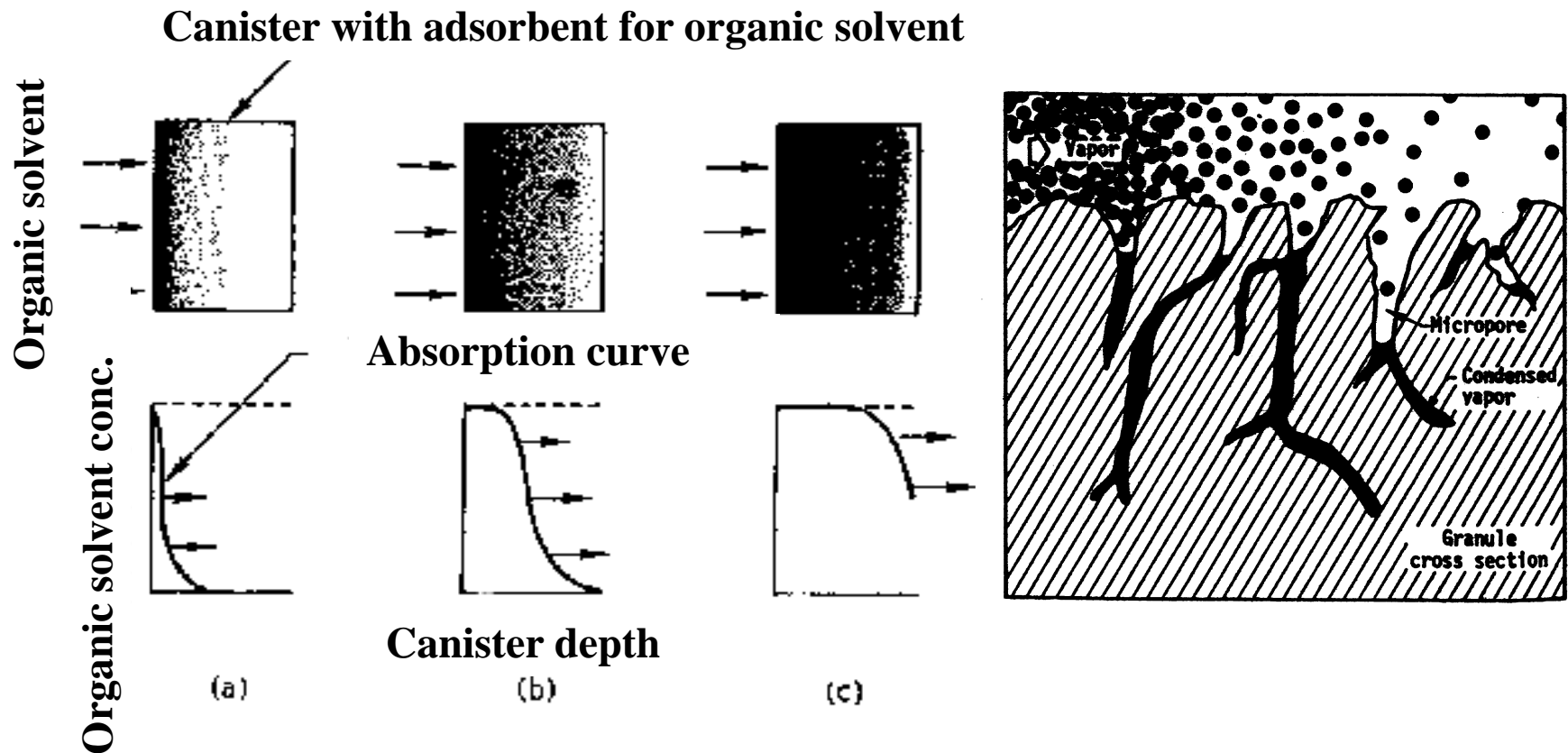
Clean fibers



Used fibers

Respirators

- “Gas masks”—removal of gaseous/vapor contaminants



Respirators

- *Particulate respirators*
(*Filtering-facepiece respirators*)



N95 half mask



N95 half mask with
exhalation valve

(Source: 3M Company, USA)

Respirators

- *Particulate respirators*
(*Filtering-facepiece respirators*)



Half facepiece with
P100 filter



Half facepiece with
P95 filter

(Source: 3M Company, USA)

Respirators

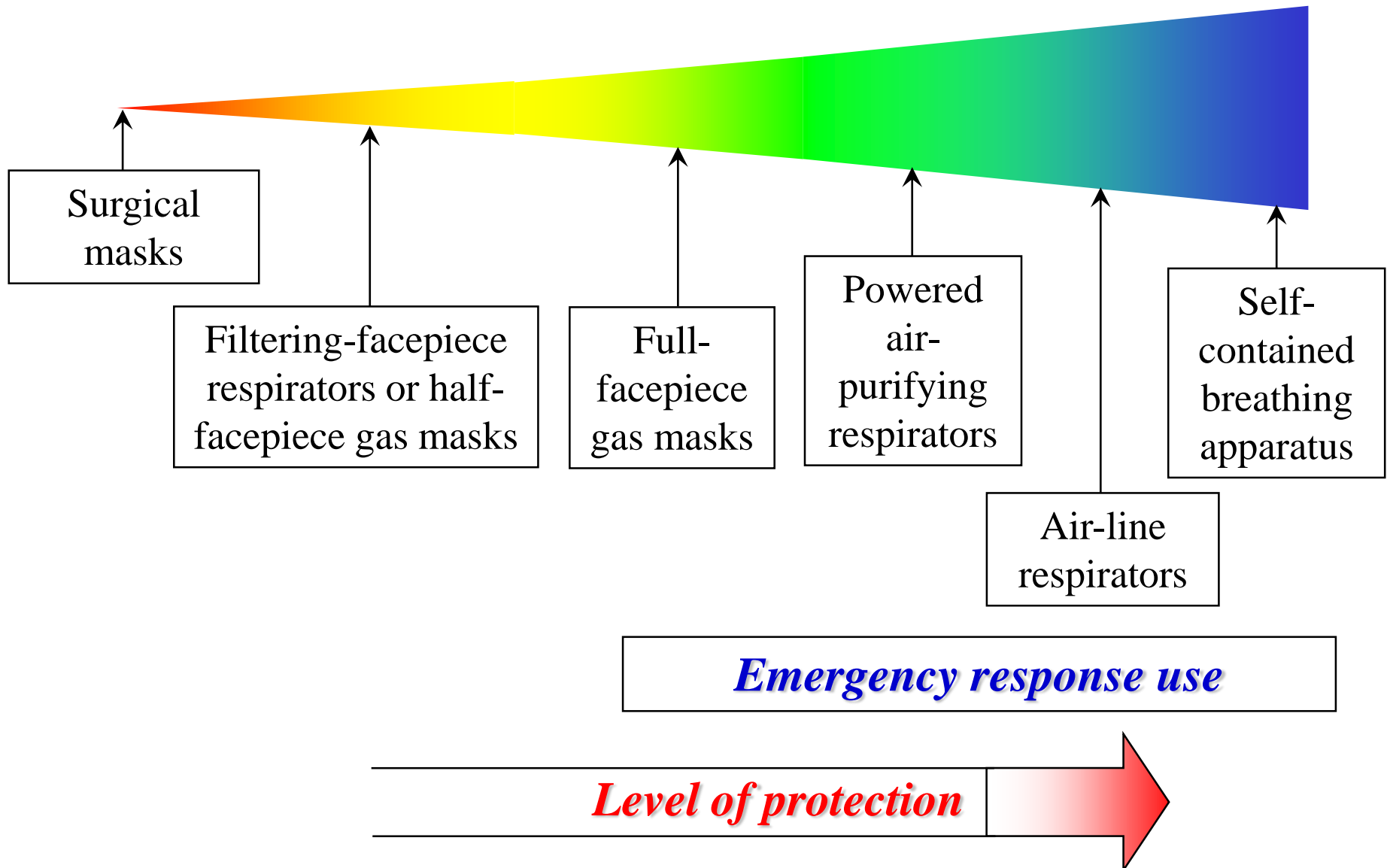
- *Gas masks (Half- and full-facepiece respirators)*
 - Facepiece and canister/cartridge combined
 - Air driven by negative pressure developed from respiration



Half-facepiece (left) and full-facepiece (right) gas mask with cartridge

(Source: 3M Company, USA)

Level of protection provided by respirators...



Classification of particulate respirators by US National Institute for Occupational Safety and Health

Series	Level	Efficiency of Filtration (%)	Targeted Contaminants	Time of Use
N	N100	99.7	Particulates and <i>non-oil-based aerosols</i>	Varied by use condition
	N99	99		
	N95	95		
R	R100	99.7	Particulates (including oil-based aerosols)	A working day (8 hours)
	R99	99		
	R95	95		
P	P100	99.7	Particulates (including oil-based aerosols)	Varied by use condition
	P99	99		
	P95	95		

Respirators

- Selection of respirators
 - 1) Type of respirator and level of protection required?
 - 2) Not everyone can use the same type of respirator
 - 3) Level of reduction in contaminant concentration by the respirator?
 - 4) “Protection factor” vs. “fit factor”

Respirators

- Presence of airborne hazard: A result of air being unsafe to breath (e.g. insufficient oxygen content) or being contaminated by a hazardous material
- Understanding IDLH (Immediately Dangerous to Life or Health concentration)—*only air-supplying respirators may be used under IDLH condition*

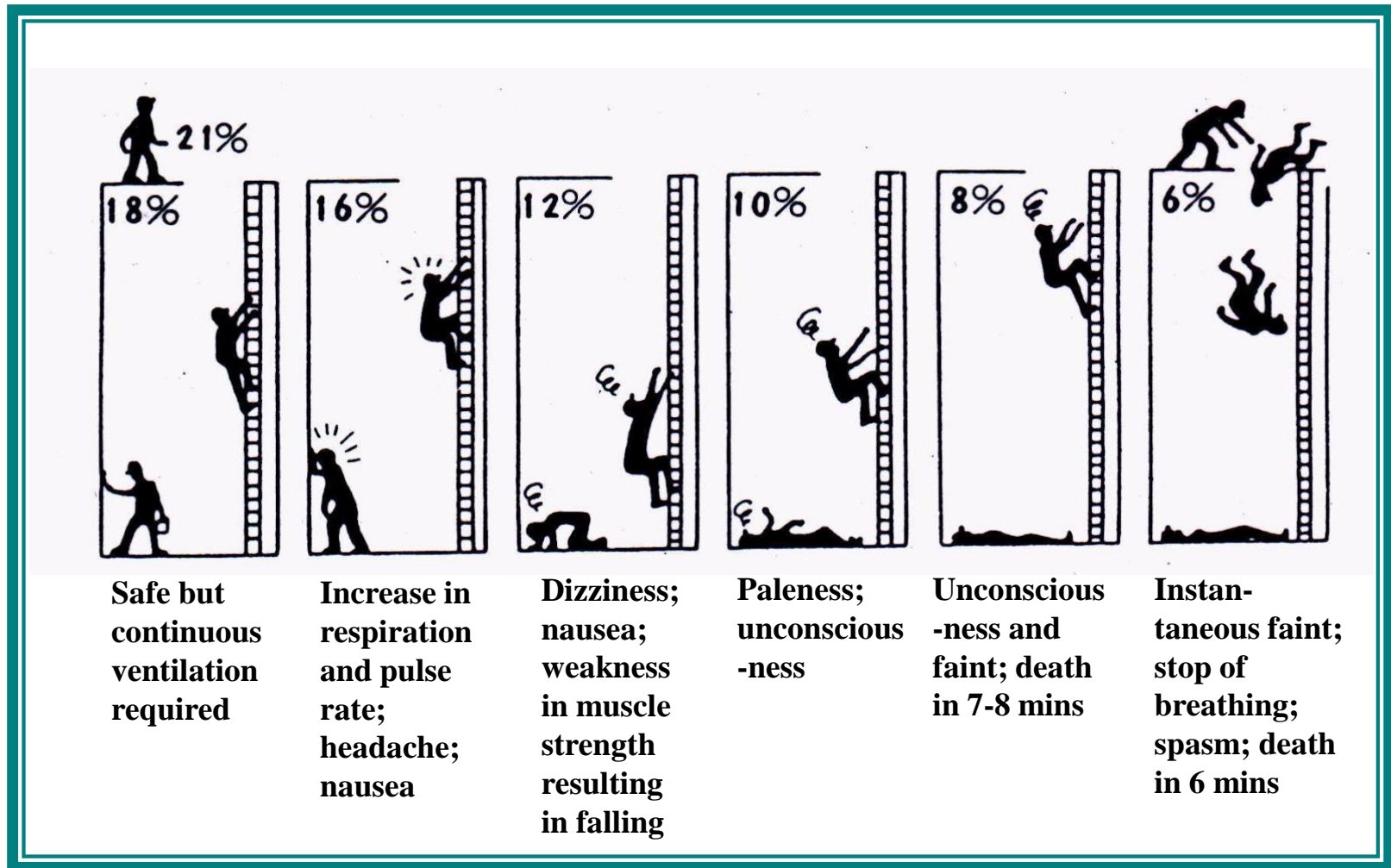


IDLH is a threshold concentration developed to prevent acute toxicity arising from inhalational exposure to hazardous material; when exposed to a target chemical at concentration beyond IDLH a person may suffer: 1) death; 2) irreversible health effect; or 3) loss of ability to escape



Respirators

- Oxygen content and physiological response



Protection Factor (PF)

$$PF = \frac{\text{Avg. conc. of contaminant in air}}{\text{Avg. conc. of contaminant inside respirator}}$$

*PF = FF (fit factor)
when filtration
efficiency reaching
100%*

Filter efficiency + Facepiece leakage

Filter efficiency testing

☞ 42 CFR Part 84

☞ EN 149:2001

☞ CNS 14755 Z2125

Fit testing

Fit Factor (PF)

- Fit of facepiece varies by model as well as by user...*

	Surgical mask 1	Surgical mask 2	Surgical mask 3	N95 mask 1	N95 mask 2
Mean	3.9	5.7	4.0	21.6	80.9
Std. Dev.	2.1	4.4	3.0	27.7	70.7
Maximum	8.4	23.3	17.0	112.6	200.0
Minimum	1.2	2.0	1.4	1.6	4.4
FF > 100	0.0%	0.0%	0.0%	6.5%	35.5%

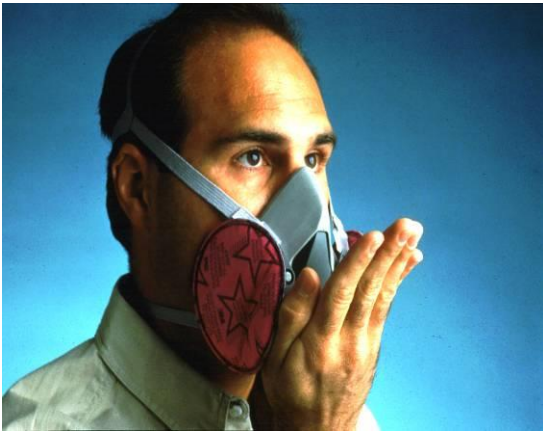
What fits you may not fit me

Respirators

- Proper use of respirators
 - 1) Select a correct and effectively working respirator
 - 2) Perform *fit check* before use
 - 3) *Fit test* (qualitative and quantitative) vs. *fit check*
(positive pressure and negative pressure)

Respirators

- Fit check ensures fit of respirator and *should be performed every time you put a respirator on*
- Demonstrated below are proper ways to perform fit check



Positive-pressure
fit check



Negative-pressure
fit check



Simplified fit check
for disposable mask

Respirators

- Fit test ensures fitting of a selected model to the user's facial characteristics and *should be performed when a new respirator is selected and also on a routine basis*



Qualitative fit test

- Dust masks: Using saccharin or Bitrex solution
- Gas masks: Using banana oil



Quantitative fit test

- Measuring concentrations present inside and outside of facepiece

Respirators

- Storage of respirators

- 1) Respirators should be prevented from

- Physical damage
- Chemical contact
- Dust contact
- Sunlight exposure
- Extreme temperature
- Excessive moisture



- 2) Respirators for emergency use should be clearly labeled with user instructions and stored in adequate, accessible places

Section 6

Chemical Protective Clothing and Hand Protection

Chemical Protective Clothing in Common Use...



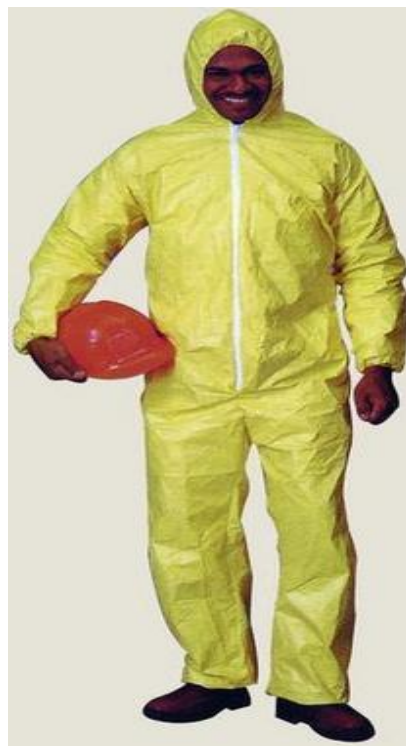
Level A Ensemble

- Airtight clothing
- Self-contained breathing apparatus (SCBA) or air-line respirator



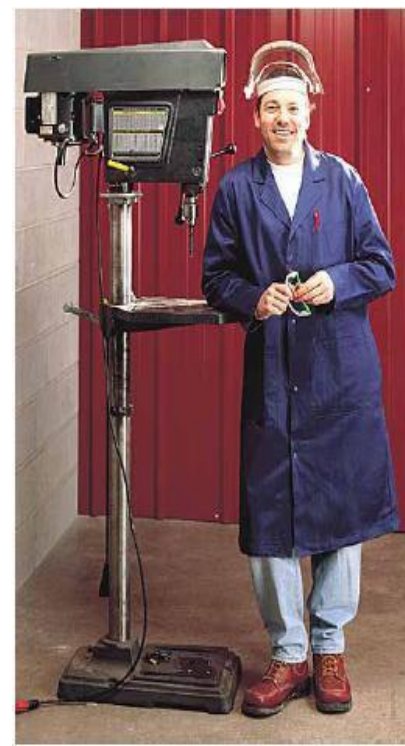
Level B

- Non-airtight clothing, gloves, boots
- SCBA or air-line respirator



Level C

- Clothing, gloves, boots
- Half-/full-facepiece respirator



Level D

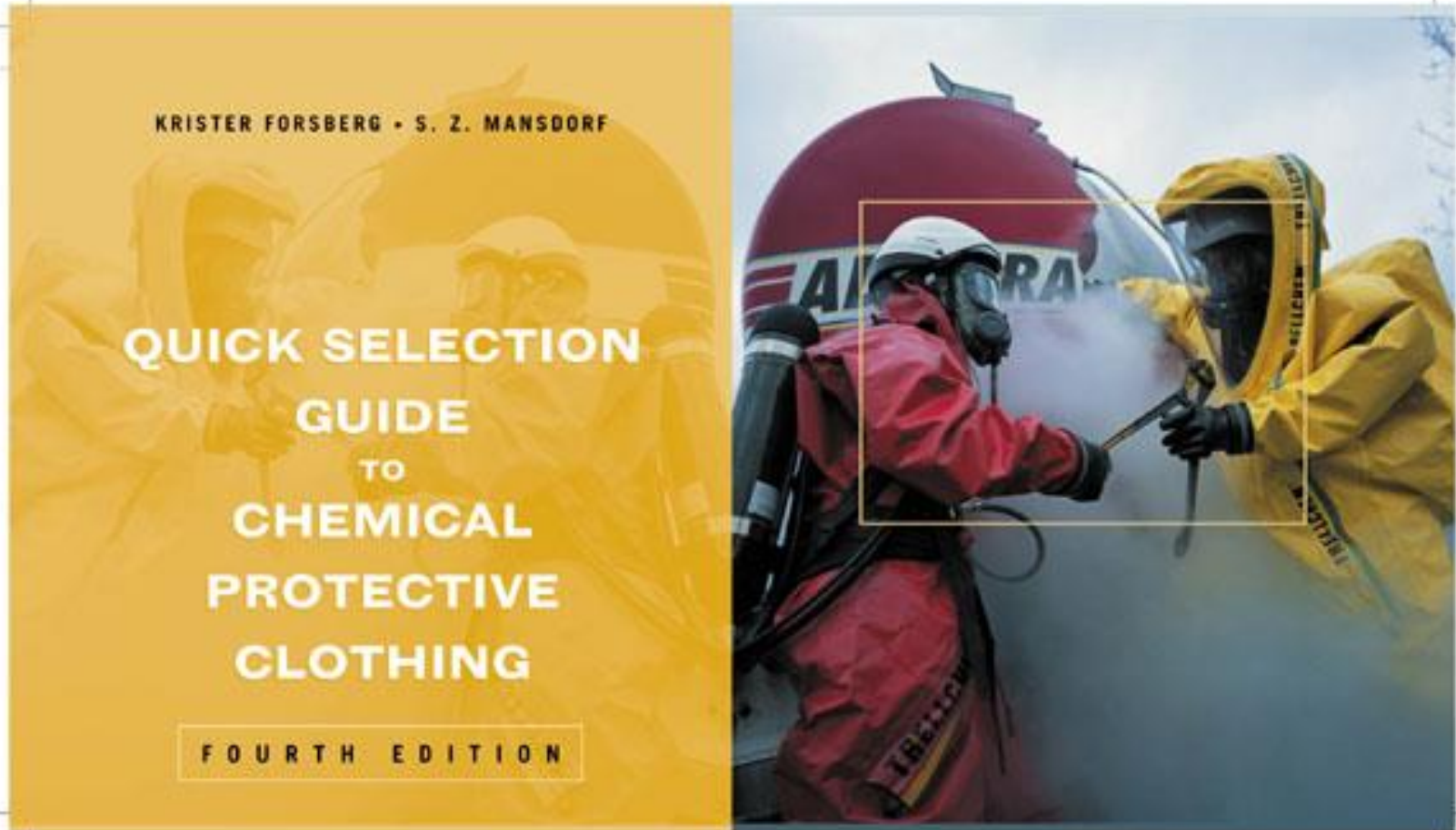
- Boots
- No respirator required

Protective Clothing

- Laboratory coat

- 1) Lab coat are required when staying in laboratory handling or storing chemical
- 2) Lab coat should be removed immediately if contaminated; white coat are preferred, and should be kept clean
- 3) Long pants and closed shoes of low heel are the preferred personal clothing; baggy clothes and long hairs are prone to accidents in a laboratory

Selection of chemical protective clothing...



(Source: Forsberg and Mansdorf, 2004)

Color-Coded Recommendations

“Green” indicates “recommended > 4 h or > 8 h”

>8	Recommended >8 h.
Green	Recommended >4 h.
Yellow	Caution 1–4 h.
Red	Not recommended <1 h. (Degradation may occur)
White	Not Tested

292 Hydrocarbons, Aromatic

Benzene
p-tert-Butyltoluene
m-Cresol
p-Cresol
 Cresols, isomeric mixture
 Cumene
 Diethylbenzene
 Divinylbenzene
 Ethylbenzene
 Gasoline, 40–55% aromatics
 Gasoline, unleaded
 Isobutylbenzene
alpha-Methylstyrene
 Styrene
 Toluene

	Butyl Rubber	Natural Rubber	Neoprene Rubber	Nitrile Rubber	Polyethylene (PE)	Polyvinylalcohol	Polyvinylchloride	Viton®	Viton®/Butyl Rubber	Barrier (PE/PA/PE)	Silver Shield/4H® (PE/EVAL/PE)	Responder®	Trellech® HPS	Tychem® BR/LV	Tychem® SL	Tychem® TK
Benzene	Red	Red	Red	Red	Red	>8	Red		>8	>8	>8	>8		>8	Red	>8
<i>p</i> -tert-Butyltoluene	Red	Red	Red	Green		>8		>8			>8			>8	Red	>8
<i>m</i> -Cresol	>8	Red	Green	Yellow	Red		Yellow	>8								
<i>p</i> -Cresol	>8	Red	>8		Red		Yellow	>8								
Cresols, isomeric mixture	>8	Red	Green	Yellow	Red		Yellow	>8				>8		>8	>8	>8
Cumene	Red	Red	Red	Yellow				>8						>8		>8
Diethylbenzene	Red	Red														
Divinylbenzene	Red	Red	Red	Red		>8	Red	>8				>8				
Ethylbenzene	Red	Red	Red		Red		Red	>8	>8			>8		>8	Red	>8
Gasoline, 40–55% aromatics	Red	Red		>8			Red	>8				>8		>8		>8
Gasoline, unleaded	Red	Red	Red	>8		Green	Red	>8	>8	Yellow				>8	Red	>8
Isobutylbenzene												>8				
<i>alpha</i> -Methylstyrene	Red	Red	Red	Red		Yellow		>8								
Styrene	Red	Red	Red	Red	Red	Green	Red	>8	>8	>8	>8	>8	>8	>8	Red	>8
Toluene	Red	Red	Red	Red	Red	>8	Red	>8		>8	>8	>8	>8	>8	Red	>8

CAUTIONS: Recommendations are NOT valid for very thin Natural Rubber, Neoprene, Nitrile, and PVC gloves (0.3 mm or less).

Hand Protection

- Death of Dr. Wetterhahn—absorption of dimethylmercury following its permeation through latex gloves
- *There is no glove material that can completely stop permeation of all chemicals; the difference lies in how long it takes for chemical to break through*



(Exposure of skin to hydrofluoric acid. Source: Taiwan Institute of Occupational Safety and Health)

Hand Protection

- Gloves commonly used in laboratory
 - Gloves made of cottons and asbestos for protection against heat and sometimes covered in alumina for protection against chemical and water
 - Plastic gloves made from mold-dipping for protection in tasks handling chemical or oil/grease
 - Gloves made of leather for protection in welding works
 - Gloves supported by metal net to prevent finger-cutting when handling knives or sharp objects

Gloves in Common Use



Natural rubber



Neoprene rubber



Nitrile rubber



Polyvinyl chloride



Polyvinyl alcohol



Butyl rubber

Rating of gloves for permeation and degradation resistance

Permeation/Degradation Resistance Guide for Ansell Gloves

The first square in each column for each glove type is color coded. This is an easy-to-read indication of how we rate this type of glove in relation to its applicability for each chemical listed. The color represents an overall rating for both degradation and permeation. The letter in each square is for Degradation alone...

- GREEN: The glove is very well suited for application with that chemical.
- YELLOW: The glove is suitable for that application under careful control of its use.
- RED: Avoid use of the glove with this chemical.



CHEMICAL	LAMINATE FILM			NITRILE			UNSUPPORTED NEOPRENE			SUPPORTED POLYVINYL ALCOHOL			POLYVINYL CHLORIDE (Vinyl)			NATURAL RUBBER			NEOPRENE/NATURAL RUBBER BLEND		
	BARRIER			SOL-VEX			29-865			PVA			SNORKEL			CANNERS AND HANDLERS*			CHEMI-PRO*		
	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate
1. Acetaldehyde	■	380	E	P	—	—	E	10	F	NR	—	—	NR	—	—	E	7	F	E	10	F
2. Acetic Acid	■	150	—	G	270	—	E	60	—	NR	—	—	F	180	—	E	110	—	E	260	—
3. Acetone	▲	>480	E	NR	—	—	E	10	F	P	—	—	NR	—	—	E	10	F	G	10	G
4. Acetonitrile	▲	>480	E	F	30	F	E	20	G	■	150	G	NR	—	—	E	4	VG	E	10	VG
5. Acrylic Acid	—	—	—	G	120	—	E	390	—	NR	—	—	NR	—	—	E	80	—	E	65	—
6. Acrylonitrile	E	>480	E	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
7. Allyl Alcohol	▲	>480	E	F	140	F	E	140	VG	P	—	—	P	60	G	E	>10	VG	E	20	VG
8. Ammonia Gas	■	19	E	▲	>480	—	▲	>480	—	—	—	—	■	6	VG	—	—	—	■	27	VG
9. Ammonium Fluoride, 40%	—	—	—	E	>360	—	E	>480	—	NR	—	—	E	>360	—	E	>360	—	E	>360	—
10. Ammonium Hydroxide	E	30	—	E	>360	—	E	250	—	NR	—	—	E	240	—	E	90	—	E	240	—

(Source: Ansell Occupational Healthcare 2003)

Hand Protection

- Indices applied in selection of gloves

- ☒ Chemical to handle

- ☒ Duration of exposure

- ☒ Material in gloves

- ☒ Dexterity and thickness

- ☒ Comfort

- ☒ Temperature at work

- ☒ Anti-aging characteristics

- ☒ Permeation rate

- ☒ Mechanical strength

- ☒ Level of voltage encountered

- ☒ Work process

- ☒ Certification of glove inspection

- ☒ Manufacturer

- ☒ Cost

Hand Protection

- Cautions on using gloves
 - The barrier material used in constructing the gloves should not allow for skin absorption of chemical or become a source of dermatoses
 - If you use barrier creams, keep in mind they do not offer a complete resistance to chemical permeation
 - When operating a machine, take the gloves off before you work; the gloves may be tangled and rolled into the operating or moving machine
 - Do not mix-use different gloves
 - The protective gloves should fit the user's hands and not interfere with capability holding objects

Hand Protection

- Inspection
 - The status of protective gloves must be checked before they are used
 - ☑ Any change in color?
 - ☑ Any puncture or sign of tearing?
 - ☑ If defect is suspected, change the gloves

Hand Protection

- Usage and limitations
 - 1) Proper donning and doffing of gloves
 - 2) Proper use of disposal vs. non-disposal gloves
 - 3) Wash hands immediately once gloves are removed
 - 4) Gloves are not to be used in tasks where drills and cutting tools are involved and finger-cutting is a risk
 - 5) Review *Material Safety Data Sheet* when organic solvent is used in task so to determine efficacy of barrier material

Hand Protection

- Steps to remove disposal gloves

1.



3.



2.



4.



Acknowledgements

- This slideshow was compiled and edited by Chen-Peng Chen of Department of Occupational Safety and Health (DOSH), China Medical University (CMU), based on the information originally presented in the following teaching materials:
 - “Management of Laboratory Safety and Health” and “Personal Protective Equipments”, Examination Center for Laboratory Safety and Health, National Taiwan University (2011)
 - “Hazard Communication for Dangerous and Harmful Materials”, Yu-Li Huang, Department of Safety, Health and Environmental Engineering, National Kaohsiung First University of Science & Technology (2010)
 - “Industrial Toxicology” and “Personal Protective Equipments”, Chen-Peng Chen, DOSH, CMU (2015)